Application No.: 10/552,507 Docket No.: 053128

AMENDMENTS TO THE CLAIMS

This listing of claims replaces all prior versions of claims in the application.

 (Currently Amended): A <u>hydrogen storage and/or transportation container comprising a</u> hydrogen storage alloy material, comprising:

wherein said hydrogen storage alloy material has a structure where ultrafine particles of Pd, M (M is at least one metal selected from the group consisting of Pt, Au, Fe, Co and Ni) and one or more compounds thereof are precipitated and dispersed in a parent phase of ZrO₂.

wherein said hydrogen storage alloy material is prepared by subjecting an amorphous Zr alloy used as a precursor to a heat treatment in air or an oxygen atmosphere so as to form the structure.

wherein the Zr alloy has a composition, in atomic %, expressed by the following formula:

 $\frac{Z_{F_100-a-b}Pd_aNi_b}{Z_{L_100-a-b}Pd_aM_b} \ \, (\text{wherein } 15 \leq a \leq 40, 2 < b \leq 10, \text{and } \underline{M} \text{ is at} \\ \, \text{least one metal selected from the group consisting of } Pt, Au, Fe, Co and Ni), wherein said hydrogen storage alloy material has a structure where said <math>Pd$, said Ni and one or more compounds thereof are dispersed in a parent phase of Z_FO_3 in the form of ultrafine particles, and wherein said hydrogen storage alloy material being prepared by subjecting an amorphous alloy to a heat treatment in air or an oxygen atmosphere.

2. (Original): The hydrogen storage alloy material as defined in claim 1, which exhibits a hydrogen storage amount of 2.5 weight % or more in a weight ratio relative to Pd contained in said hydrogen storage alloy material. Application No.: 10/552,507 Docket No.: 053128

- 3. (Cancelled).
- (Currently Amended): A method for producing the hydrogen storage alloy material as defined in claim 1, comprising:

preparing a melt of a master Z_L alloy formed through a melting process; rapidly solidifying said melt at a cooling rate of 10^4 K/s or more to form said amorphous Z_L alloy; and

subjecting said amorphous \underline{Zr} alloy to an oxidizing heat treatment in air or an oxygen atmosphere at 250 to 350°C to selectively oxidize said alloy element Zr so as to allow <u>ultrafine</u> <u>particles of said Pd</u>, said \underline{Ni} and one or more compounds thereof to be <u>precipitated and</u> dispersed in a parent phase of ZrO_2 in the form of nanoparticle-size ultrafine particles.